Compound Quality in Medicinal Chemistry: Considerations Beyond Molecular Properties

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Monitoring and improving the quality of the compounds screened in drug discovery projects is a concept familiar to all medicinal chemists. The term 'quality' here implicitly refers to an assessment of the molecular properties of the compound and how they relate to enhancing the likelihood of that compound becoming a drug (e.g. by application of Lipinski's rules or related analyses). We propose that, whilst valid, this may be overlooking something more fundamental. There is an equally important consideration; the concept of 'quality' should also refer to the provenance of the compound, i.e. how (a) it was selected for synthesis, (b) that process was carried out and (c) a successful outcome was verified. We suggest that these simple-sounding operations, although apparently obvious to any chemist, may sometimes be overlooked. It is fundamentally important to both consider and evidence all aspects relating to compound quality, in any project or process.

1. CONTEXT:

Compounds tested in biological assays are usually the products of a chemical reaction. By definition, each reaction has inputs (reagents, starting materials) and outputs (desired products, side products, reagent by-products) along with solvents, catalysts, reaction vessels etc., which facilitate the desired transformation, but are intended (or expected) to remain unchanged. Process chemists are well accustomed to consideration of these aspects, but chemists working on earlier stage projects (Hit-to-Lead, Lead Optimisation) may pay less attention to apparently obvious (but potentially problematic) variables.

2. DATA OVERLOAD:

Chemists have access to many tools to facilitate compound design and synthesis. They select what (in their opinion) are the best starting materials (e.g. most cost / time effective) along with the reaction conditions which seem likely to effectively deliver the desired product(s). Huge repositories of publicly available analytical and characterisation data exist to facilitate comparisons for identification purposes (e.g. CAS references > 190 million organic & inorganic substances, ZINC15: > 230 million compounds, e-Molecules: 24.6 million screening compounds and 9 million building blocks). Selecting the best compound inputs (I) and suitable chemical reactions (R) to give a particular compound output (O), and ensuring that all information, individual data points and parameters used are fit-for-purpose is critical. However, the huge size of this data set may present an 'information overload' challenge for chemists and encourage assumptions to be made, thereby hindering effective decision making and influencing the outcome.



3. SELECTION BIAS:

The assumptions (or bias) made may be reasonable, or ignoring them may have no significant consequence. However this may not always be the case. For example, a structurally incorrect building block ('input') may be selected, or the chemical reaction used may be sub-optimal, for reasons which are not immediately obvious. Chemists usually anticipate this type of random error occurring from time-to-time; it's put down to experience, and they use one of several possible back-up options to ensure success in obtaining the desired compound output. However, the situation is potentially more complex than this, and all the potential assumptions may not always be factored into the decision-making process. Negative consequences can result e.g. wasted time, financial loss and potential pursuit of the wrong direction or lead series for a project. Worst case scenarios could include problems with late-stage compounds, issues with data reproducibility or retractions of papers.

		Variable	Potential consequence(s)	Possible solution(s)	4 SOLUTION:	
	Input	Is the reagent supplied as stated on the bottle label? Could it have been labelled wrongly, or become contaminated by a previous user?	 Failed reaction Wrong product prepared / isolated Safety issue 	 Check data sheet Independent analysis at point of receipt / use 	We propose that generating data (evidence) to confirm assumptions made (or bias) should be undertaken more frequently and we should rely on 'data not dogma', even if the answer may seem obvious. The potential impact of being wrong may have detrimental consequences for a project. A wide range of variables and associated assumptions exist – each with multiple potential pitfalls, but usually having simple solutions to mitigate them. A non-exhaustive selection of these variables, common assumptions made, potential consequences and possible solutions are provided in the Table, categorised into Input / Reaction / Output, along with publications of relevance to each. This list is intended to provide some food for thought and illustrate common assumptions that can be made, or where bias can arise during the synthetic process.	
		Isomers (regio, geometric, chirality etc) of 1 the reagent are as stated on the bottle / vial	 Failed reaction Wrong product prepared / isolated 	 Independent analysis at point of receipt / use 		
		How long has the bottle of reagent sat on the shelf in the storeroom, and under what conditions?	Failed reactionUnexpected low yield	Stock room logsIndependent analysis pre-use		
	Reactions	Are the reaction conditions described ² in a paper as producing 85% yield of product correct, or was it a 'one-off' best case?	Lower yield than statedLoss of starting materialFailed reaction	 Adopt same approach as biology with 'n=3' technical replicates 		
		Are the components stable under 3,4 stated conditions?	Failed reactionFailure to isolate desired product	Careful monitoring		
	Output	Has structure been assigned accurately by paper / patent authors and/ or has all necessary analytical data been used and interpreted correctly?	 Wrong compound tested Inaccurate SAR Project direction changed 	 Independent confirmation of structure / regiochemistry using in-house data with no inherent assumptions 		
		Is the isolated product a free base, salt, hydrate or combination thereof (despite what may be stated in a procedure)	 Unexpected physicochemical properties Incorrect stoichiometry calculations 	Enhanced analysis, (e.g.) elemental / combustion		
		How sure can one be that there are no undesired, low-level and potentially problem-causing impurities in any 'purified' and isolated sample(s)?	 Failed reaction 'Surprise' reaction(s) Unexpected or erroneous screening results 	Use enhanced purification and more stringent / targeted analysis		
	AVOID ASSUMPTIONS! Key variables to actively consider before, during and after any synthetic procedure:INPUT: verify reagent ID and qualityREACTION: verify procedure suitability and information qualityOUTPUT: verify product ID and quality					
 Check your chemistry, Baker, Nature 2017, 485–488; Chirality: a key parameter in chemical probes, McGown et al., RSC Chem. Biol. 2023, 716-721 On the Practical Limits of Determining Isolated Product Yields and Ratios of Stereoisomers: Reflections, Analysis, and Redemption, Wemerova et al., Synlett, 2010, 2701–2707 Investigating the Underappreciated Hydrolytic Instability of 1,8-Diazabicyclo[5.4.0]undec-7-ene and Related Unsaturated Nitrogenous Bases, Hyde et al., Org. Process Res. Dev. 2019, 1860–1871 Chemical Instability and Promiscuity of Arylmethylidenepyrazolinone-Based MDMX Inhibitors, Stefaniak et al., ACS Chem. Biol. 2018, 2849–2854 The Dimroth rearrangement as a probable cause for structural misassignments in imidazo[1,2-a]pyrimidines: A 15N-labelling study and an easy method for the determination of regiochemistry, Chatzopoulou et al., Tetrahedron, 2018, 5280–5288 						
o. Filot Study to Quantity Panadium Impurities in Lead-like Compounds Following Commonly Used Purification Techniques, Chatzopoulou & Madden et al., ACS Med. Chem. Lett. 2022, 262–270; Guideline for Analysis and Preventio Contamination Catalysis, Daru et al., Angew. Chem. Int. Ed., 2025, e202424425						

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